

# Can Pb-free Halide Double Perovskites Support High-efficiency Solar Cells?

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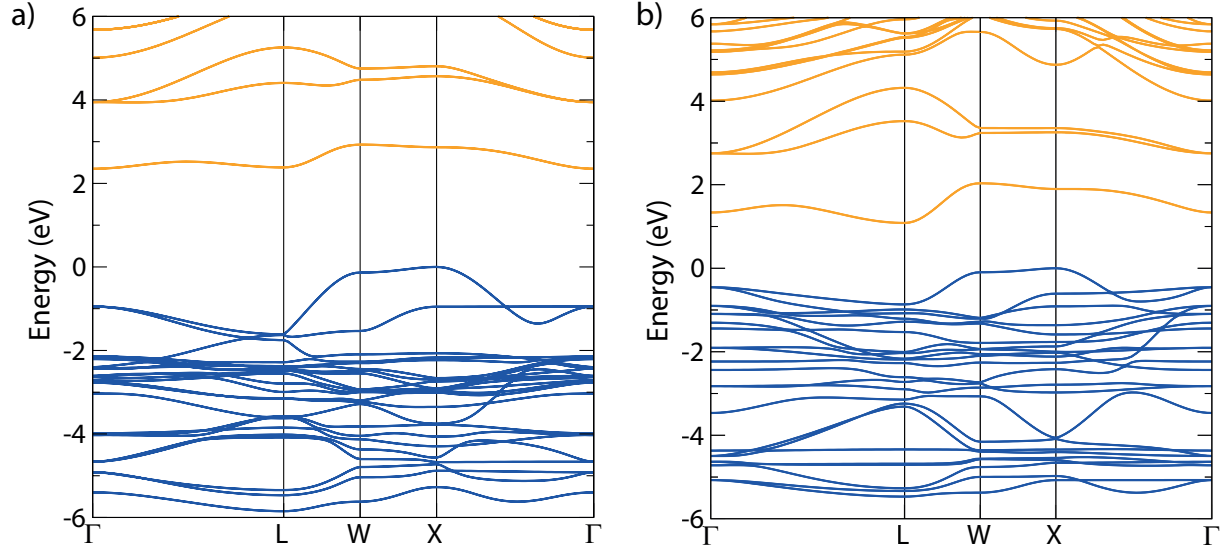


Figure 1: HSE06+SOC band structures of a)  $\text{Cs}_2\text{AgBiCl}_6$  and b)  $\text{Cs}_2\text{AgBiI}_6$ , with valence band in blue, and conduction band in orange. Valence band maximum (VBM) is set to 0 eV.

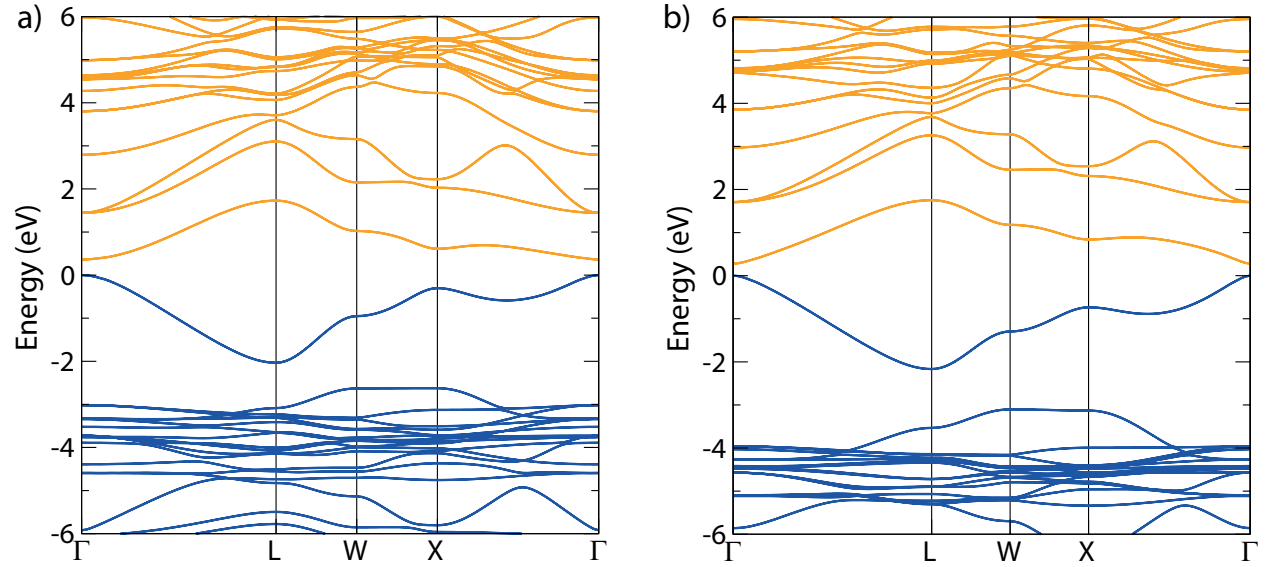


Figure 2: HSE06+SOC band structures of a)  $\text{Cs}_2\text{InBiCl}_6$  and b)  $\text{Cs}_2\text{InBiBr}_6$ , with valence band in blue, and conduction band in orange. VBM is set to 0 eV.

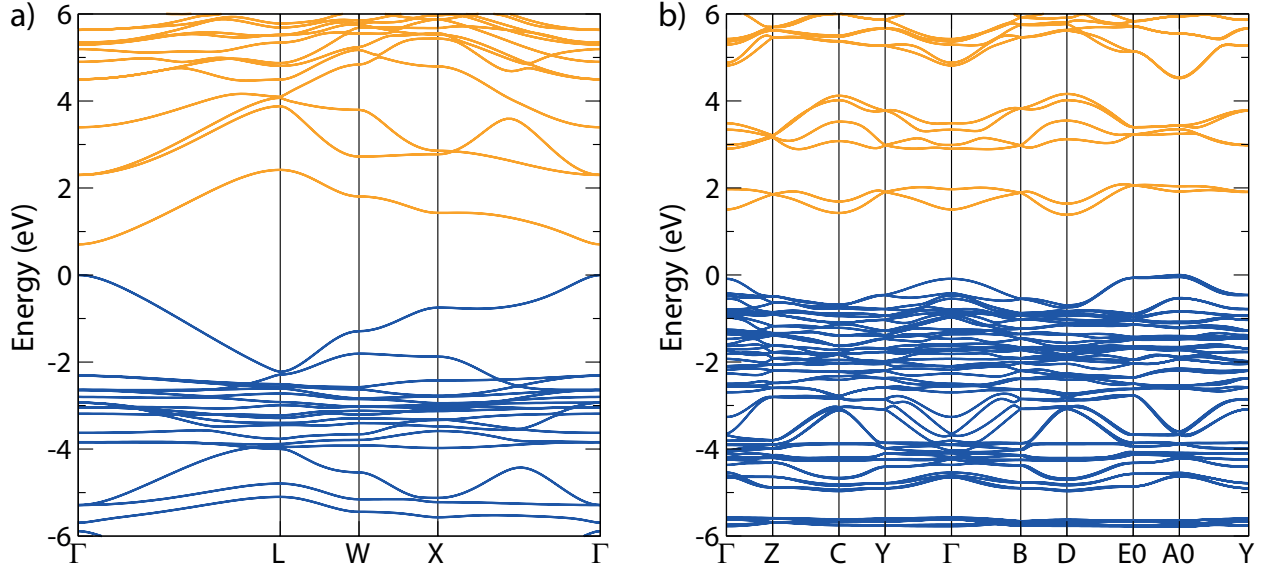


Figure 3: HSE06+SOC band structures of a)  $\text{Cs}_2\text{TlBiBr}_6$  and b)  $(\text{CH}_3\text{NH}_3)_2\text{AgBiI}_6$  (starting from the orthorhombic  $(\text{CH}_3\text{NH}_3)\text{PbI}_3$  structure)<sup>1</sup> with valence band in blue, and conduction band in orange. The high-symmetry path used for  $(\text{CH}_3\text{NH}_3)_2\text{AgBiI}_6$  uses k-points from the lowered C-centered monoclinic space group. VBM is set to 0 eV.

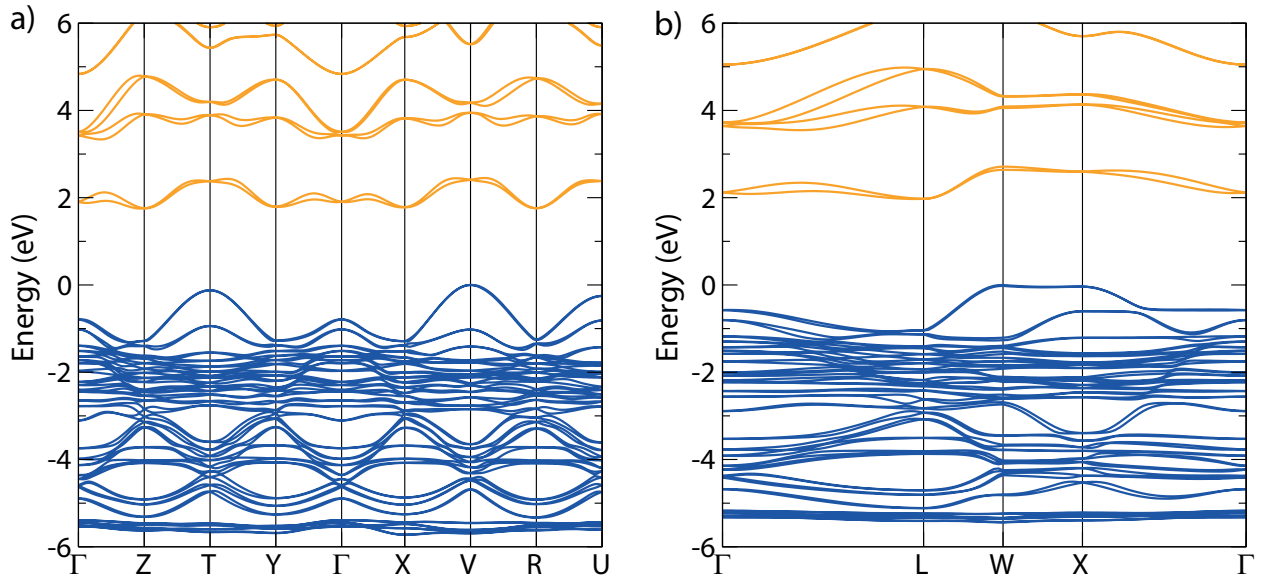


Figure 4: HSE06+SOC band structures of  $(\text{CH}_3\text{NH}_3)_2\text{AgBiBr}_6$  (starting from the primitive cell of the cubic  $\text{Cs}_2\text{AgBiBr}_6$  structure),<sup>2</sup> with valence band in blue, and conduction band in orange. a) uses the high-symmetry path of the lowered primitive triclinic space group, while b) uses the F-centered cubic high-symmetry lines. VBM is set to 0 eV.

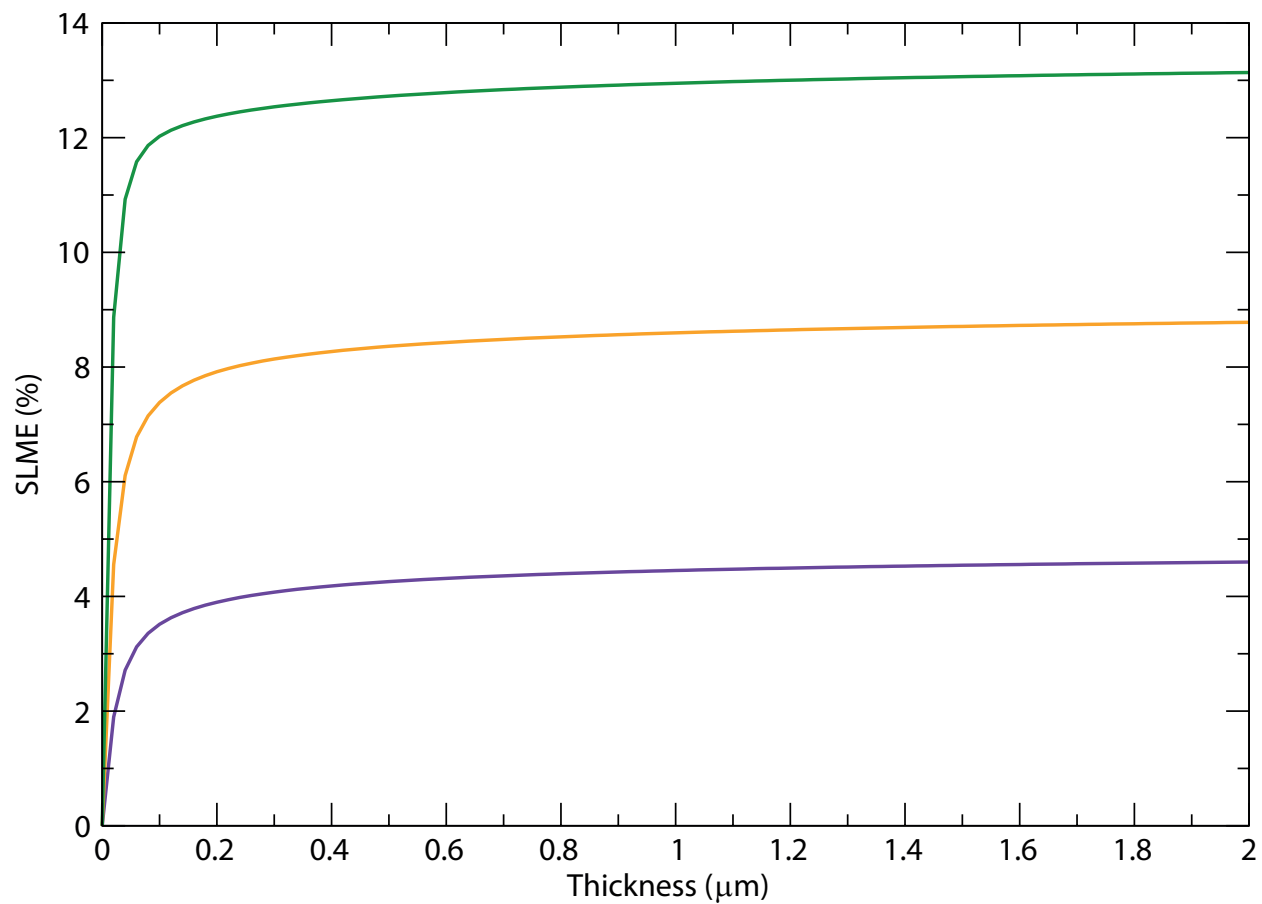


Figure 5: Plot of the predicted SLMEs against film thickness for  $\text{Cs}_2\text{AgBiCl}_6$  (green),  $\text{Cs}_2\text{AgBiBr}_6$  (orange) and  $\text{Cs}_2\text{AgBiI}_6$  (purple).

## References

- (1) Weller, M. T.; Weber, O. J.; Henry, P. F.; Pumpo, M. D.; Hansen, T. C. Complete structure and cation orientation in the perovskite photovoltaic methylammonium lead iodide between 100 and 352 K. *Chemical Communications* **2015**, *51*, 4180–4183.
- (2) McClure, E. T.; Ball, M. R.; Windl, W.; Woodward, P. M. Cs<sub>2</sub>AgBiX<sub>6</sub> (X = Br, Cl): New Visible Light Absorbing, Lead-Free Halide Perovskite Semiconductors. *Chemistry of Materials* **2016**, *28*, 1348–1354.